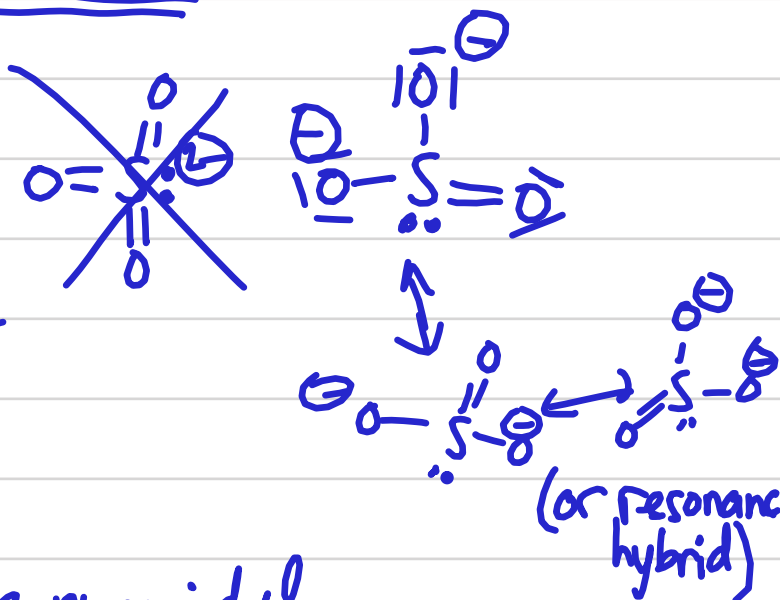


Chem 151B, Spring 2018
Problem Set #1

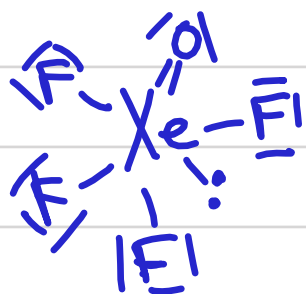


$$6 + 18 + 2 = 26e^-$$

(Trigonal) pyramidal



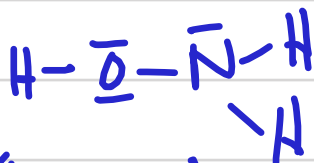
$$8 + 6 + 28 = 42e^-$$



square pyramidal



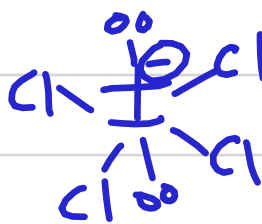
$$1 + 6 + 5 + 2 = 14e^-$$



(Trigonal) pyramidal



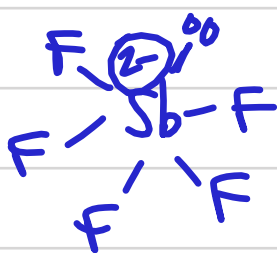
$$7 + 28 + 1 = 36e^-$$



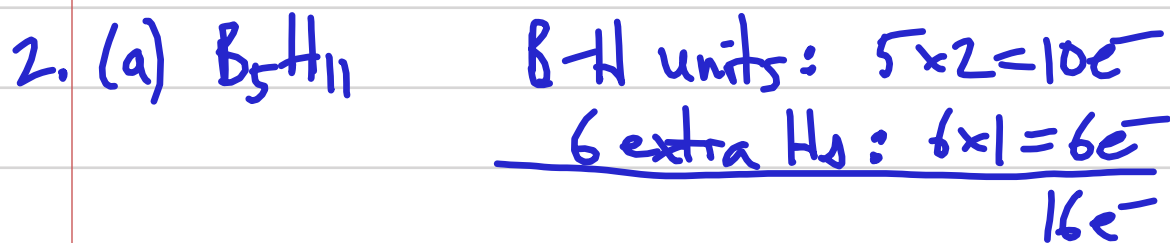
sq. planar



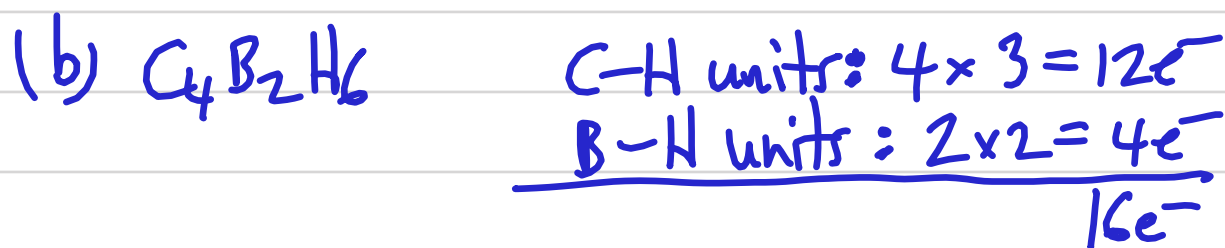
$$5 + 35 + 2 = 42e^-$$



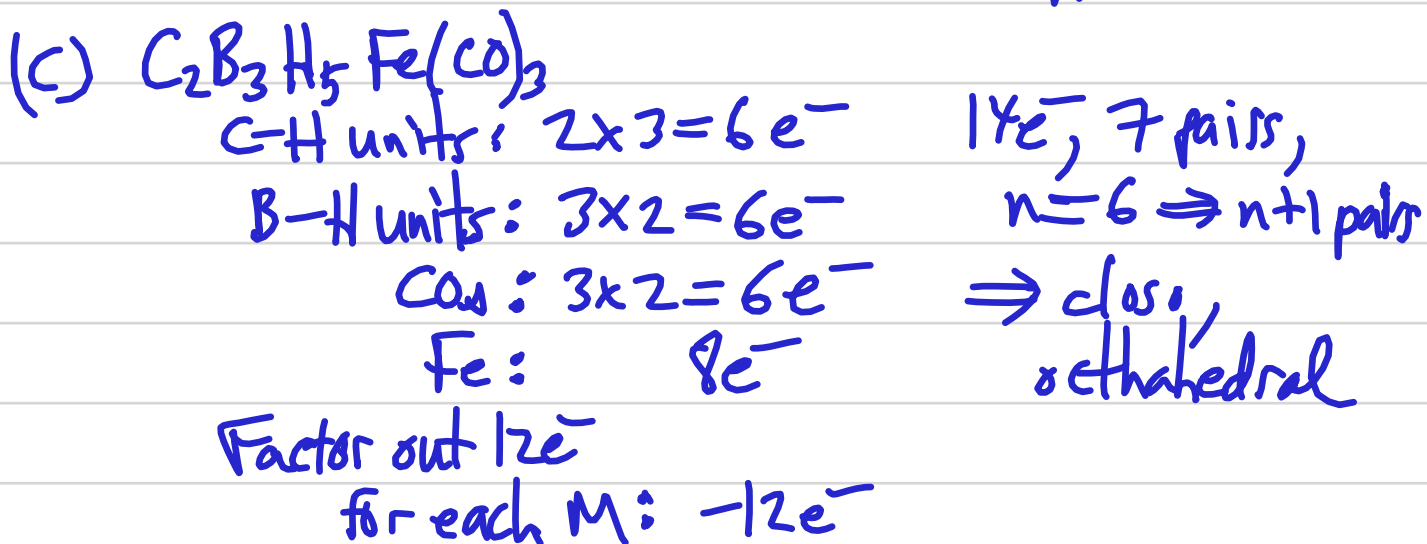
sq. pyramidal

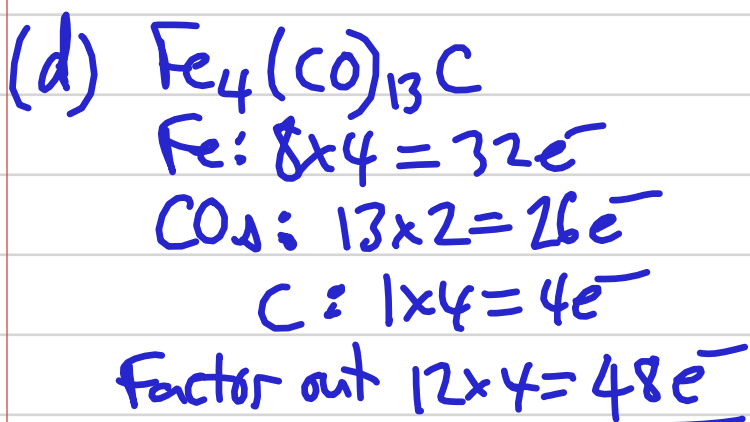


8 e^- skeletal e^- pairs, $n=5 \Rightarrow n+3$,
 arachno, based on $5+2=7$ vertex polyhedron,
 missing two vertices
 PBP missing two vertices (one axial, one
 equatorial)
 (enough for answer) ("pentagonal" also ok)



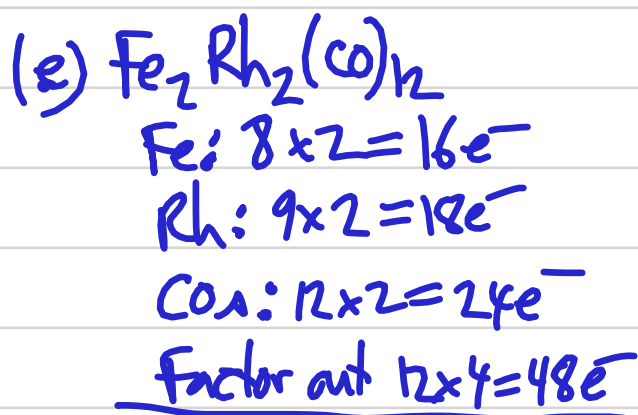
8 skeletal e^- pairs, $n=6 \Rightarrow n+2$,
 nido, based on $(n+1)=7$ -vertex polyhedron
 PBP missing one vertex or pentagonal
 pyramidal





$14e^-$

7 pairs, $n=4$ (C in center)
 $\Rightarrow n+3$, arachno, based on
 6-vertex polyhedron,
 octahedral, missing two vertices,
 see-saw/disphenoidal
 (sq. pl. ok)



$10e^-$

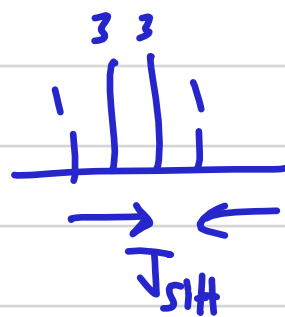
5 pairs, $n=4$
 $\Rightarrow n+1$ pairs,
 closo,
 tetrahedral

3. $\text{SiH}_3\text{Cl}(g) \leftarrow \Rightarrow$ isotropic

^{29}Si , $I = \frac{1}{2}$, $N = 4.7\%$

^1H , $I = \frac{1}{2}$, $N = 99.9\%$

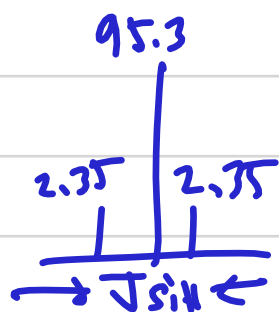
^{29}Si :

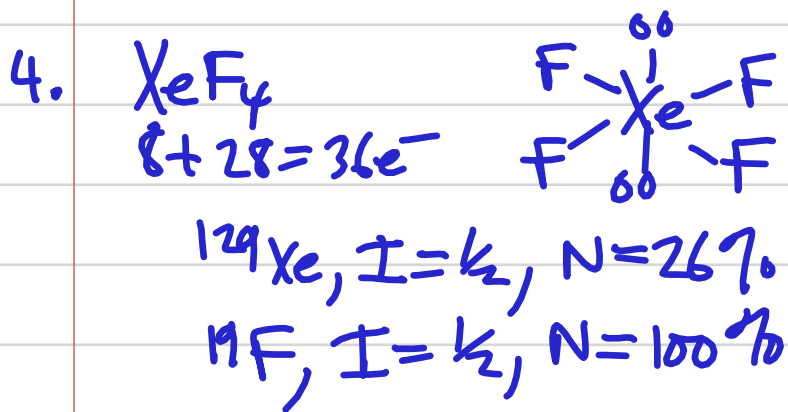


^1H : C_{3v} , all Hs equivalent

95.3% bonded to ^{28}Si , no splitting

4.7% bonded to ^{29}Si , $2nI+1 = 2(1)(\frac{1}{2})+1 = 2$, doublet





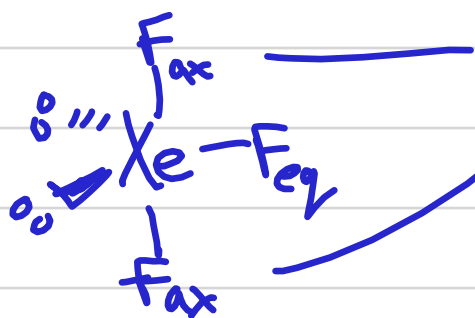
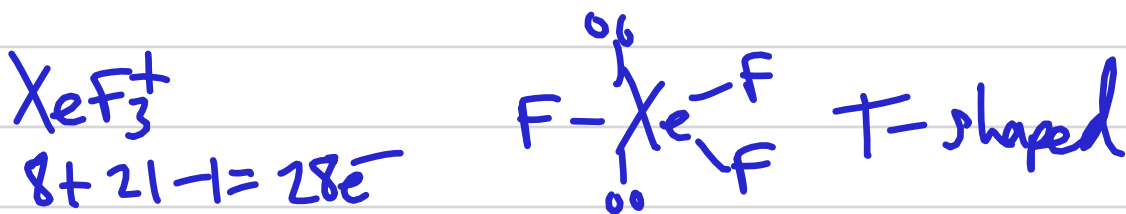
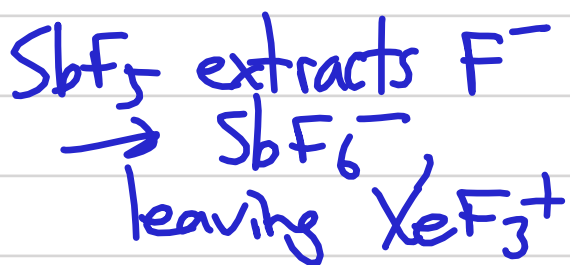
sq. planar

^{129}Xe NMR:
 couples to 4 equivalent
 F_s

$$2nI + 1 = 2(4)(\frac{1}{2}) + 1$$

$$= 5$$

1:4:6:4:1 quintet



Axials split, $2nI + 1 = 2(2)(\frac{1}{2}) + 1 = 3$

1:2:1 triplet

E_q split, $2nI + 1 = 2(1)(\frac{1}{2}) + 1 = 2$

1:1 doublet

1:2:1 triplet of 1:1 doublets



$J_{Xe-F_{eq}} > J_{Xe-F_{ax}}$
 ∴ question said triplet,
then each component
 split further
 into doublet

(i.e. not a doublet of triplets)